Phosphinoarsonium and Diarsonium Cations from Chloride-Abstraction-Induced Pnictogen Coupling

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Received February 15, 2008

The first salts of phosphinoarsonium, diarsonium, and phosphinodiarsonium ions are obtained from reaction mixtures of a chlorophosphine, a tertiary arsine, and a chloride-abstracting agent. The versatile synthetic approach has implications in the development of a wide variety of interpnictogen compounds.

A variety of high-yield preparative reactions have been reported that provide a catalog of *catena*-phosphorus cations.^{1–10} Together with the neutral^{11–14} and anionic^{11–13,15–24} analogues, *catena*-phosphorus compounds have the potential to achieve the extent and diversity of chemistry based on *catena*-carbon frameworks. The prototypical phosphinophosphonium **1**, ⁴ diphosphonium **4**, ²⁵ and phosphinodiphospho-

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nium $7¹$ ions as well as larger frameworks^{5–10} have been substitutionally derivatized; however, diversification within the pnictogen skeleton is currently limited to derivatives of framework 2 (Pn = As,²⁶ Sb, Bi²⁷). Triphosphorus cations of type **7** were first obtained from reaction mixtures of a dichlorophosphine, a tertiary phosphine, and aluminum $chloride¹$ and we have now applied this approach more generally to prepare the first salts of phosphinoarsonium **3**, diarsonium **6**, and phosphinodiarsonium **9** ions.

Inorg. Chem. **²⁰⁰⁸**, *⁴⁷*, 2952-²⁹⁵⁴

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 $31P$ NMR spectra of reaction mixtures composed of ClPR₂ $(R = Me, Ph)$, Me₃As, and Me₃SiOSO₂CF₃ (TMSOTf) in

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2952 Inorganic Chemistry, Vol. 47, No. 8, 2008 10.1021/ic800287e CCC: \$40.75 [©] 2008 American Chemical Society Published on Web 03/15/2008

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Table 1. 31P NMR Data for Derivatives of **2**, 26,28,29 **3**, and **9**

Figure 1. Representative solid-state views of the cation in [**3**(Ph)][OTf] (**I**), [**6**][OTf]2 (**II**), and [**9**(Me)][OTf]2 (**III**). Ellipsoids are at 50% probability. Hydrogen atoms not shown.

 $CH₂Cl₂$ show the formation of a single phosphorus-containing species that has been isolated and comprehensively characterized as the trifluoromethanesulfonate (triflate) salt of phosphinoarsonium cations $3(R)$ ($R = Me$, Ph). The analogous reaction of Cl_2PR ($R = Me$, Et, Ph, Cy, ⁱPr), Me₃As, and TMSOTf gives
salts of phosphinodiarsonium cations **9**. In comparison, the salts of phosphinodiarsonium cations **9**. In comparison, the combination of Cl3P with Me3As and TMSOTf gives hexamethyldiarsonium triflate [6][OTf]₂. Table 1 lists ³¹P NMR data for derivatives of **3** and **9**, together with values for[**2a**]- $[PF_6]$,²⁸ [2b][BPh₄],²⁶ and [2c][OTf].²⁹ X-ray crystallographic data confirm the identity of [**3**(Ph)][OTf], [**6**][OTf]2, and [**9**(Me)][OTf]2, and representative views of the cations in the solid state are shown in Figure 1. Selected structural parameters for the cations are listed in Table 2, together with comparative values for derivatives of the arsinophosphonium framework (**2**).26,28,29

Chlorophosphines (ClPR₂, Cl₂PR) are activated by TMSOTf, which effects chloride abstraction and in the presence of the tertiary arsine Me3As enables the formation of coordinate $P \leftarrow$ As bonds in **3** and **9**, more appropriately

described as a phosphinoarsonium bond. The consequential pnictogen coupling is a rational and high-yield synthetic approach to interpnictogen frameworks. The formation of the diarsonium cation 6 from Cl₃P, Me₃As, and TMSOTf represents a redox process that is induced by chloride abstraction and implicates the formation of phosphorus subchlorides or elemental phosphorus. This reaction contrasts those of Cl₃P, R₃P, and AlCl₃, which give $[R_3PPPR_3]$ - $[AlCl₄]^{2,3}$ We are currently targeting derivatives of framework **5** through alkylation of framework **3** and framework **8** through reactions of dichloroarsines with tertiary phosphines in the presence of a halide-abstracting agent.

31P NMR chemical shifts for derivatives of **3** and **9** occur in a narrow range that is typical of phosphine centers in phosphinophosphonium cations.4 Indeed, the chemical shifts of the phosphonium centers in derivatives of **2** are only slightly more positive.^{26,28,29} The As-P bond lengths in **3**(Ph) and derivatives of **9** are very consistent and are slightly shorter than those in derivatives of arsinophosphonium **2** cations. The pyramidal environments for the phosphorus centers in **3**(Ph) and **9**(Me), with angles of close to 100°, are typical of phosphine centers, and the As-P-As angles in 9 (Me) are predictably larger than $As-P-C$ angles because of steric interactions between the terminal substituents on arsenic. For the same reasons, the arsonium centers in **3**(Ph) and in **9**(Me) are also substantially distorted from a tetrahedral geometry. The eclipsed conformation of the diarsonium cation **6** reveals an As-As bond length $[2.4109(3)$ Å] that is only slightly shorter than that in a neutral diarsine $([{\rm MesC(O)}]_2$ As-As[C(O)Mes]₂ = 2.4300(8) Å).³⁰ For comparison, a larger difference is observed between the Sb-Sb bond length in the hexamethyldistibonium dication [2.7624(11) and $2.7867(12)$ Å and in tetramethyldistibine $[2.862(2),$ 2.830(1), and 2.838(1) \AA ³¹

In summary, reaction mixtures of chlorophosphines with Me₃As and TMSOTf result in P-As bond formation via chloride abstraction to give the first examples of phosphi-

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noarsonium cations. Similarly, dichlorophosphines give the first phosphinodiarsonium dications. In contrast, trichlorophosphine oxidizes Me3As to give the first diarsonium dication. This versatile synthetic approach has implications in the development of a wide variety of interpnictogen compounds.

Acknowledgment. We thank the Natural Sciences and Engineering Research Council of Canada, the Canada Research Chairs Program, the Canada Foundation for Innovation, and the Nova Scotia Research and Innovation Trust Fund for funding, Dr. J. J. Weigand for helpful discussions, and the Atlantic Region Magnetic Resonance Centre for use of instrumentation.

Supporting Information Available: All experimental details including crystallographic information files (CIF) for all of the compounds presented above. This material is available free of charge via the Internet at http://pubs.acs.org.

IC800287E